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Carbene complexes: XXII. Preparation, properties and structures of the *N*,*N*-functionalized bis(amino)carbenemolybdenum(O) carbonyls [Mo(CO)₄ {

 $CN(CH_2CH:CH_2)(CH_2)_2N\}(CH_2CH:CH_2)\}]$ and



 $Mo(CO)_3 \{CN[(CH_2)_3PPh_2] (CH_2)_2N \{(CH_2)_3PPh_2\}]_n \Leftrightarrow$

Dedicated to Professor G.P. Chiusoli, a respected colleague, in recognition of his many significant contributions to organometallic chemistry, on the occasion of his 70th birthday.

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Abstract

Treatment of 1,2-bis(γ -diphenylphosphinopropylamino)ethane [obtained from 1,2-bis(allylamino)ethane and 2PPh₂H and CH(NMe₂)(OMe)₂] yielded the electron-rich enetetramine \square CN(R)(CH₂)₂NR]₂ [R \square CH₂)₃PPh₂] [abbreviated as $L_2^{(CH_2)_3PPh_2}$]. Although L_2^{All} , the analogue in which R \square CH₂CH:CH₂, is unknown, the carbenemolybdenum title compounds [Mo(CO)₄(L^{All}] (3) and \square Mo(CO)₃{L^{(CH₂)₃PPh₂]_n (4) have been prepared from [Mo(CO)₆]: for 3 from CH(NMe₂)(OMe)₂ + [All(H)N(CH₂)]₂ and for 4 from $L_2^{(CH_2)_3PPh_2}$. In 3 only one of the R groups behaves in a ligating mode. L^{All} being a chelating ligand, 3 is thus a rare example of a metal complex having both carbene- and η^2 -alkene-ligation and the first for molybdenum. The presumed polymeric structure of 4 is believed to arise from one of the P^{III} sites behaving in a chelating fashion with the other in a bridging mode. Treatment of 3 with ¹³CO stereoselectively gave the isotopomer in which a CO *trans* to a CO was displaced, the *trans*}

effect thus being parallel to the *trans* influence (X-ray data); a coproduct was *cis*-[Mo(CO)₄(13 CO) (L^{All})]. Similarly, *cis*-[Mo(CO)₄(L^{All})(PEt₃)] was the sole product of reaction between **3** and PEt₃, which upon irradiation yielded *fac*- together with *mer*-[Mo CO)₃(L^{All})(PEt₃)]. Thermolysis of **3** afforded AllC: N(CH₂)₂NAll as the only organic product. The X-ray structure of **3** showed that the mutually *trans* Mo \square CO bonds are significantly longer [av. 2.037(5) Å] than Mo \square CO *trans* to carbene [1.984(4) Å] or alkene [1.975(5) Å], with Mo \square C_{carb} at 2.229(4) Å.



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