

Nuclear Medicine



**RADIOPHARMACEUTICAL  
CHEMISTRY**

# An alpha particle consists of two protons and two neutrons

- **Common alpha-particle emitters**
  - Radon-222 gas in the environment
  - Uranium-234 and -238) in the environment
  - Polonium-210 in tobacco

# Radiation Types - Beta



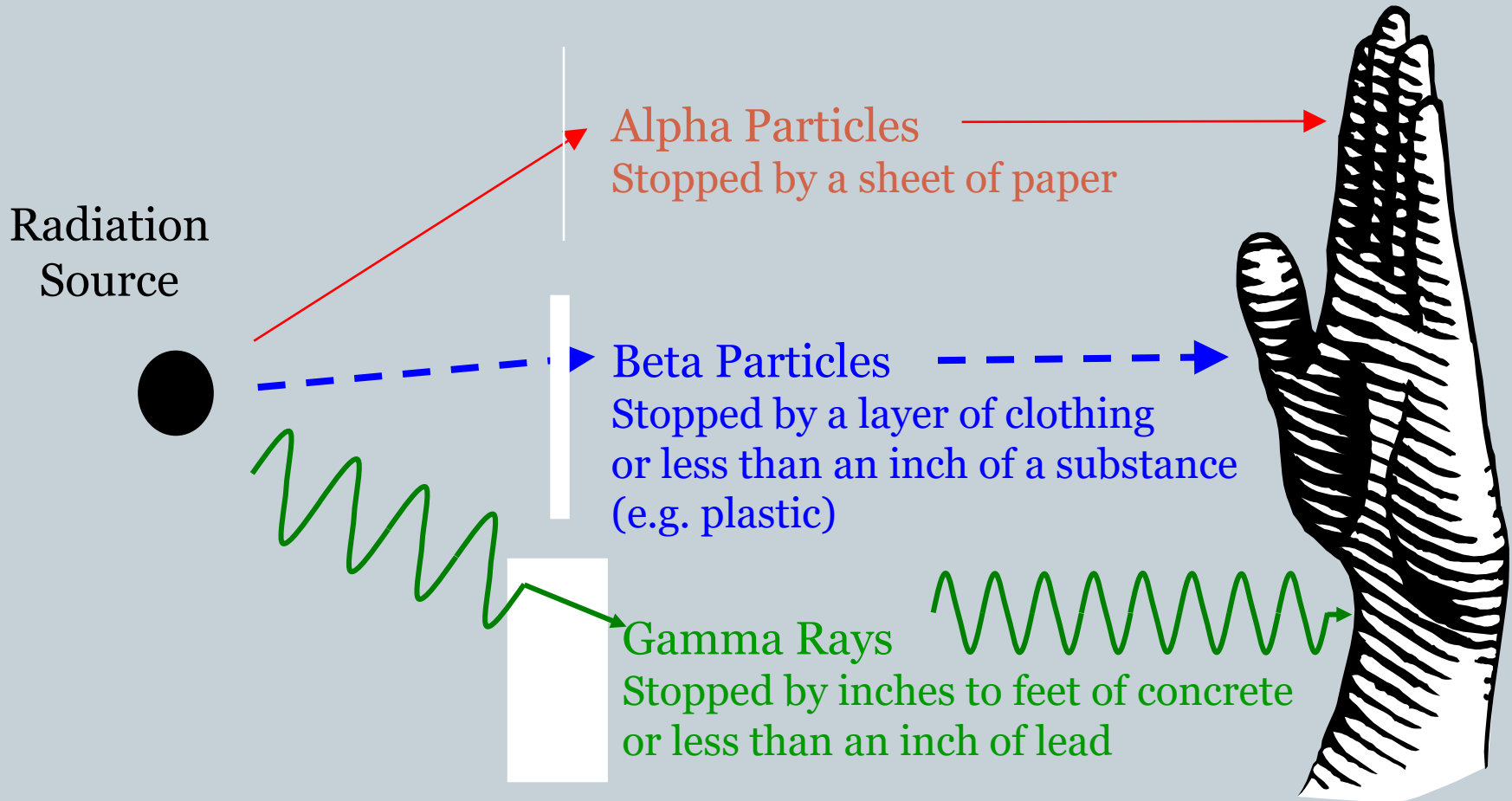
- **A beta particle is a charged electron**
  - Has the size and weight of an electron
  - Can be positively or negatively charged
- **Penetration in materials**
  - At low energies, a beta particle is not very penetrating – stopped by the outer layer of skin or a piece of paper
  - At higher energies, a beta particle may penetrate to the live layer of skin
- **Common beta-particle emitters**
  - Tritium (hydrogen-3) in the environment
  - Carbon (14) in the environment
  - Phosphorus (32) used in research and medicine

# Radiation Types - Photon



- A photon is an x or gamma ray
  - Has no weight
  - Has no charge
- Penetration in materials
  - At low energies, a photon can be stopped by a very thin (almost flexible) layer of lead or several centimeters of tissue
- At higher energies, inches of lead might be necessary to stop a photon and they can pass right through a human
- Common photon emitters
  - Cesium (137)
  - Technetium (99m) used in medicine
  - Iodine (131) used in medicine

# Types of Ionizing Radiation



Radiation  
Source

Alpha Particles  
Stopped by a sheet of paper

Beta Particles  
Stopped by a layer of clothing  
or less than an inch of a substance  
(e.g. plastic)

Gamma Rays  
Stopped by inches to feet of concrete  
or less than an inch of lead

$\alpha$  Radiation

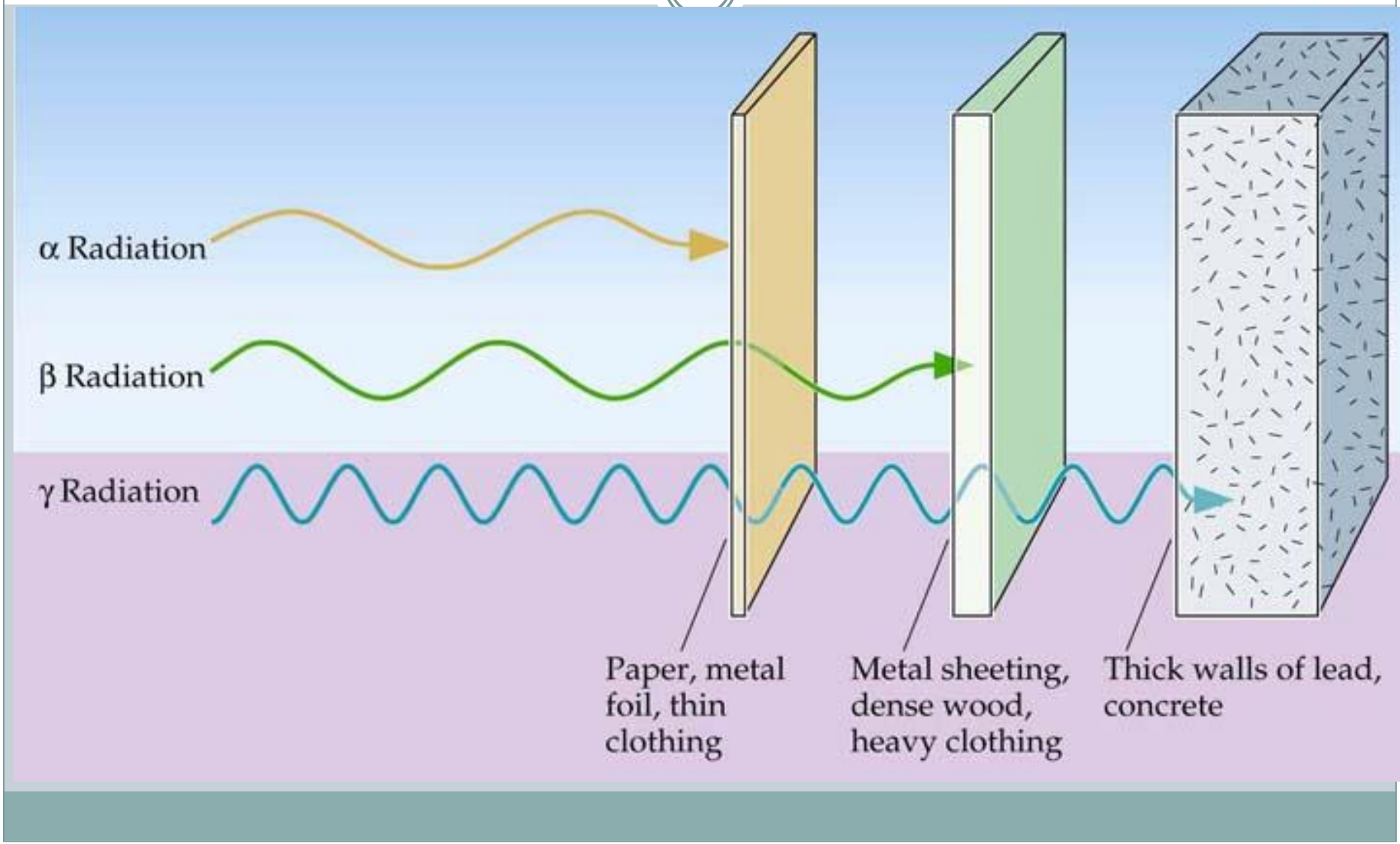
$\beta$  Radiation

$\gamma$  Radiation

Paper, metal foil, thin clothing

Metal sheeting, dense wood, heavy clothing

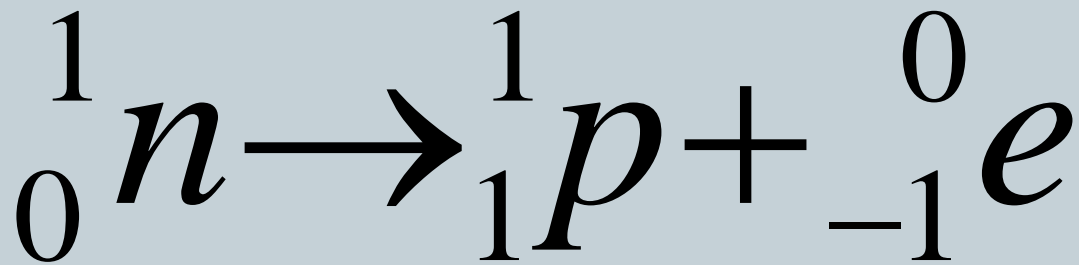
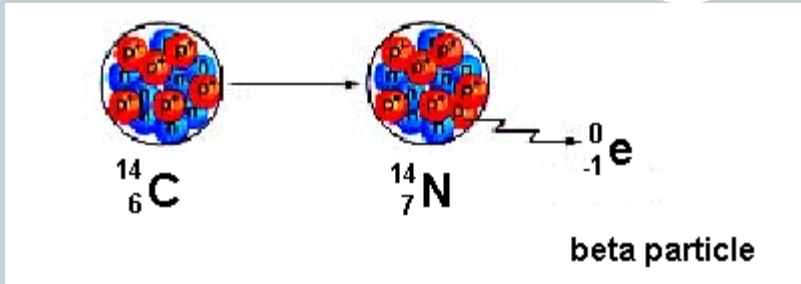
Thick walls of lead, concrete



# Products of Natural Radioactivity



Particle*	Symbol	Charge	Mass	
			Number	Identity
Alpha	${}^4_2\alpha$	2+	4	Helium
Beta	${}^0_{-1}\beta$	1-	0	Electron
Gamma light	${}^0\gamma$	0	0	Proton of





# Induced Nuclear Reactions



# Clinical Uses

## Therapeutically

**ROUTE OF ADMINISTRATIONS:**

**TELE THERAPY**

**IMPLANTATION BY WIRE OR CAPSULE**



**DIAGNOSTICALLY**

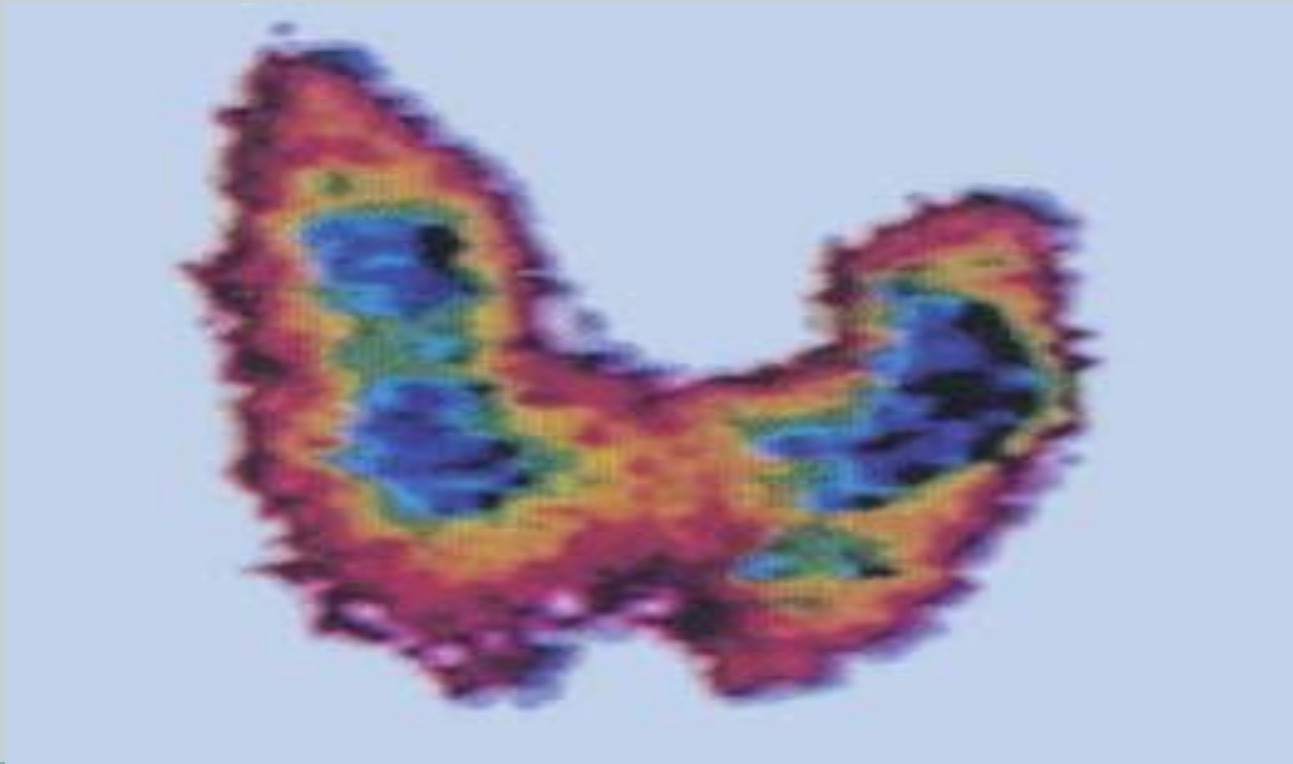
**ROUTE OF ADMINISTRATIONS:**

**ORALLY**

**PARENTERAL(IV)**

**INHALE**

An image of a thyroid gland obtained through the use of radioactive iodine.



# Positron Emission Tomography



- ❑ What is PET ?
- ❑ By which principle it work ?
- ❑ How can be performed ?
- ❑ Benefits and risks ?
- ❑ Common isotopes used?
- ❑ Main differences between PET and CT scan ?

PET scan is a functional imaging technique that is used to observe metabolic processes in the body.



**Positron emission tomography (PET):**

uses small amounts of radioactive materials called radiotracers (the biological active molecule chosen Fludeoxyglucose (FDG) an analogue of glucose)

a special camera and a computer to help evaluate your organ and tissue functions.

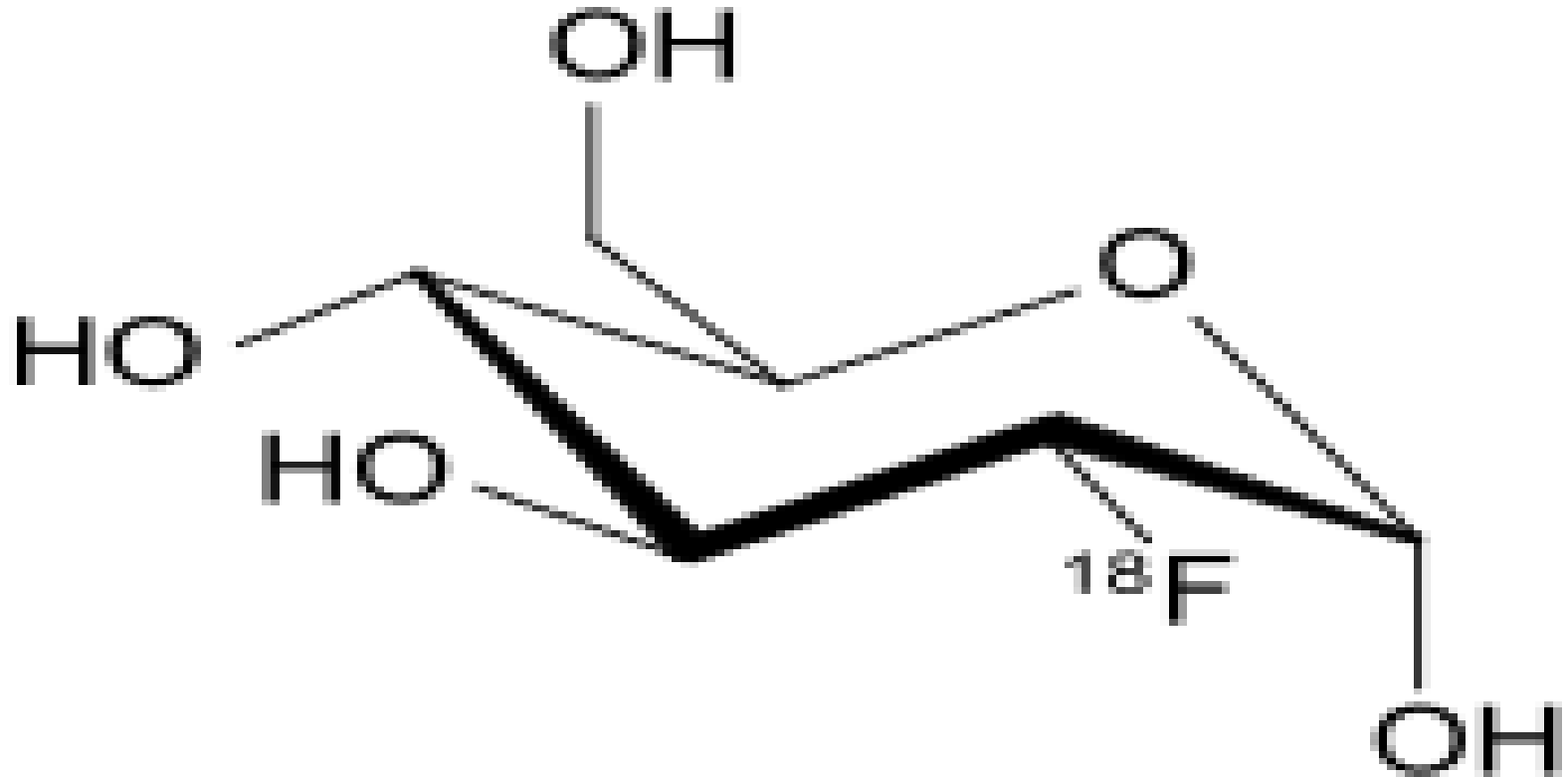
By identifying body changes at the cellular level, recently use of PET to detect the cancer tumours, may also be used in cardiology and neurology.

# PET principle



- PET use x-ray in order to present a 3D picture
- short line radioactive tracer isotope injected into the living subject usually in blood circulation ,the tracer is chemically incorporated into a biological active molecule
- There is a waiting period while the active molecule become concentrated in tissues of inert
- Radioisotope undergo a PE decay it will emit a positron

# Fludeoxyglucose(FDG) an analogue of glucose



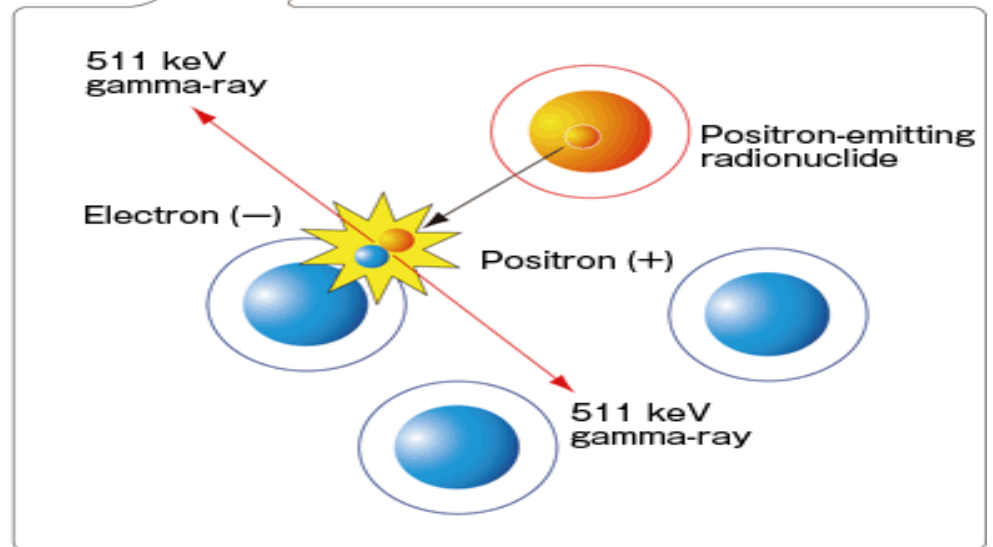
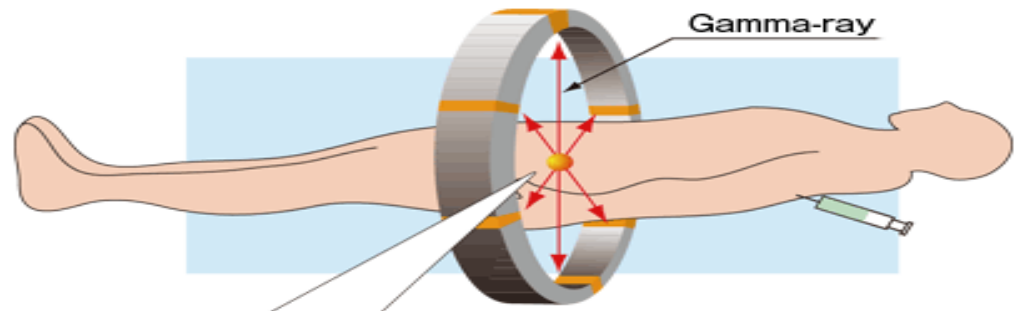
# PET principle



- Travelling few(milli meter)the positron encounter an electron
- The encounter annihilates them both produce a pair of (gamma) photon moving in opposite directions
- These are detected when they reach scintillators in the scanning device create a burst of light
- The technicians can the create an image of the parts of brain for example which are overactive



# Principle of work



# How can we performed it ?



- If you're pregnant or believe you could be pregnant, tell your doctor. The test may be unsafe for your baby. You should also tell your doctor about any medical conditions you have. If you have diabetes, you'll get special instructions for test preparation because fasting beforehand could affect your blood sugar levels.

# How can we performed it ?



- Before the scan, you'll get tracers through a vein in your arm, through a solution you drink, or in a gas you inhale. Your body needs time to absorb the tracers, so you'll wait about an hour before the scan begins.
- Next, you'll undergo the scan. This involves lying on a narrow table attached to a PET machine, which looks like a giant letter "O." The table glides slowly into the machine so that the scan can be conducted.

# How can we performed it ?



- You'll need to lie still during the scan
- You may be asked to hold your breath for short periods.
- You'll hear buzzing and clicking noises during the test.

When all the necessary images have been recorded, you will slide out of the machine. The test is then complete.

A picture clarify the process of diagnosis of pathological conditions or any other abnormalities and complications ,



**Performance of PET scan**

# Benefits



- medicine examinations provide unique information—including details on both function and anatomic structure of the body .
- nuclear medicine scans yield the most useful information needed to make a diagnosis or to determine appropriate treatment .
- Nuclear medicine is less expensive and may yield more information than surgery by identifying changes in the body at the cellular level .

# Risks

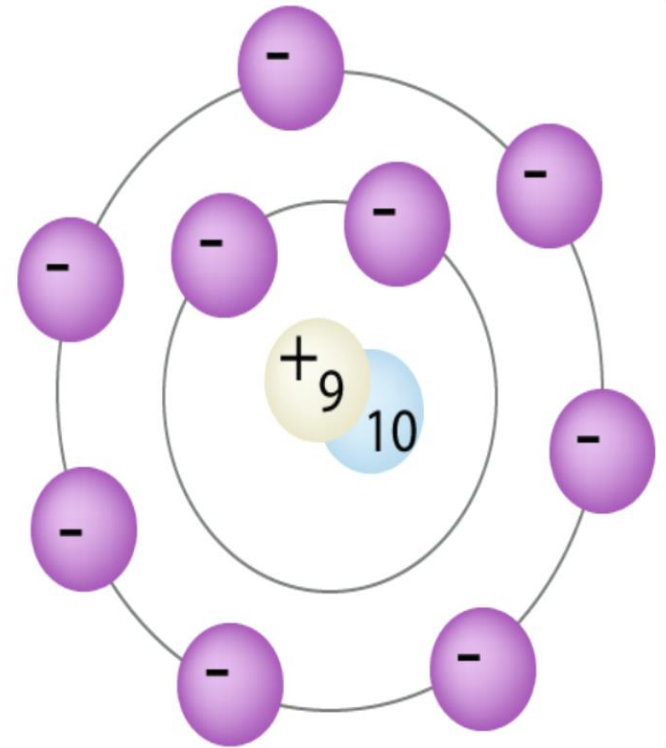
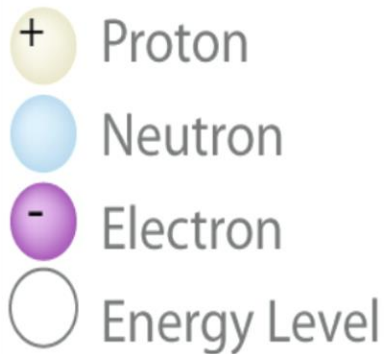


A radioactive drug (tracer) will be put into body. The amount of radiation was exposed, and the risk of negative effects

1. Cause a major allergic reaction.
2. Expose unborn baby to radiation if pregnant
3. Expose your child to radiation if breast-feeding

Isotope:- each of two or more forms of the same element that contain equal numbers of protons but different numbers of neutrons in their nuclei, and hence differ in relative atomic mass but not in chemical properties

## Fluorine (F)



**Fluorine isotope  
(most common use isotope )**



# Common isotops used in PET scan



Isotope	Half-life	Maximum energy (Mev)	Range in water (mm)
$^{18}\text{F}$	109.7 min	0.635	2.39
$^{11}\text{C}$	20.4 min	0.96	4.11
$^{13}\text{N}$	9.96 min	1.19	5.39
$^{15}\text{O}$	2.07 min	1.72	8.2

# Difference between PET and CT scan



## PET

- 1) Higher cost than CT
- 2) Take 2-4 hours to complete scan
- 3) Radiation exposure from moderate to high
- 4) Use radioactive tracer
- 5) Can image biological process of the body

## CT scan

- 1) Cost less than PET and MRI
- 2) Completed within 5 min ,so its less sensitive to patient
- 3) The effective radiation dose from CT ranges from 2 to 10 (mSv)
- 4) Use x-ray for imaging
- 5) CT can outline bone inside the body very accurately

# Difference



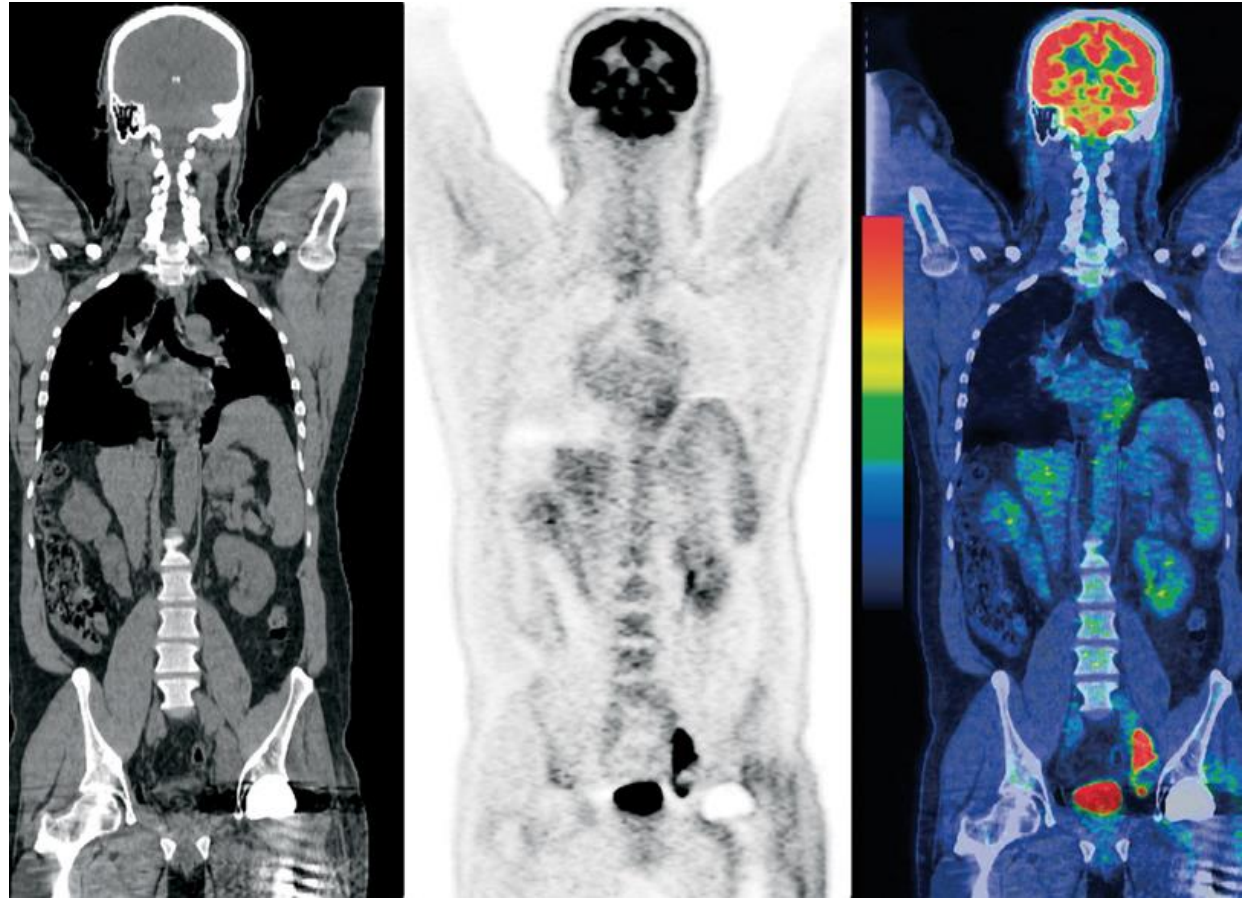
**CT scan**



**PET scan**

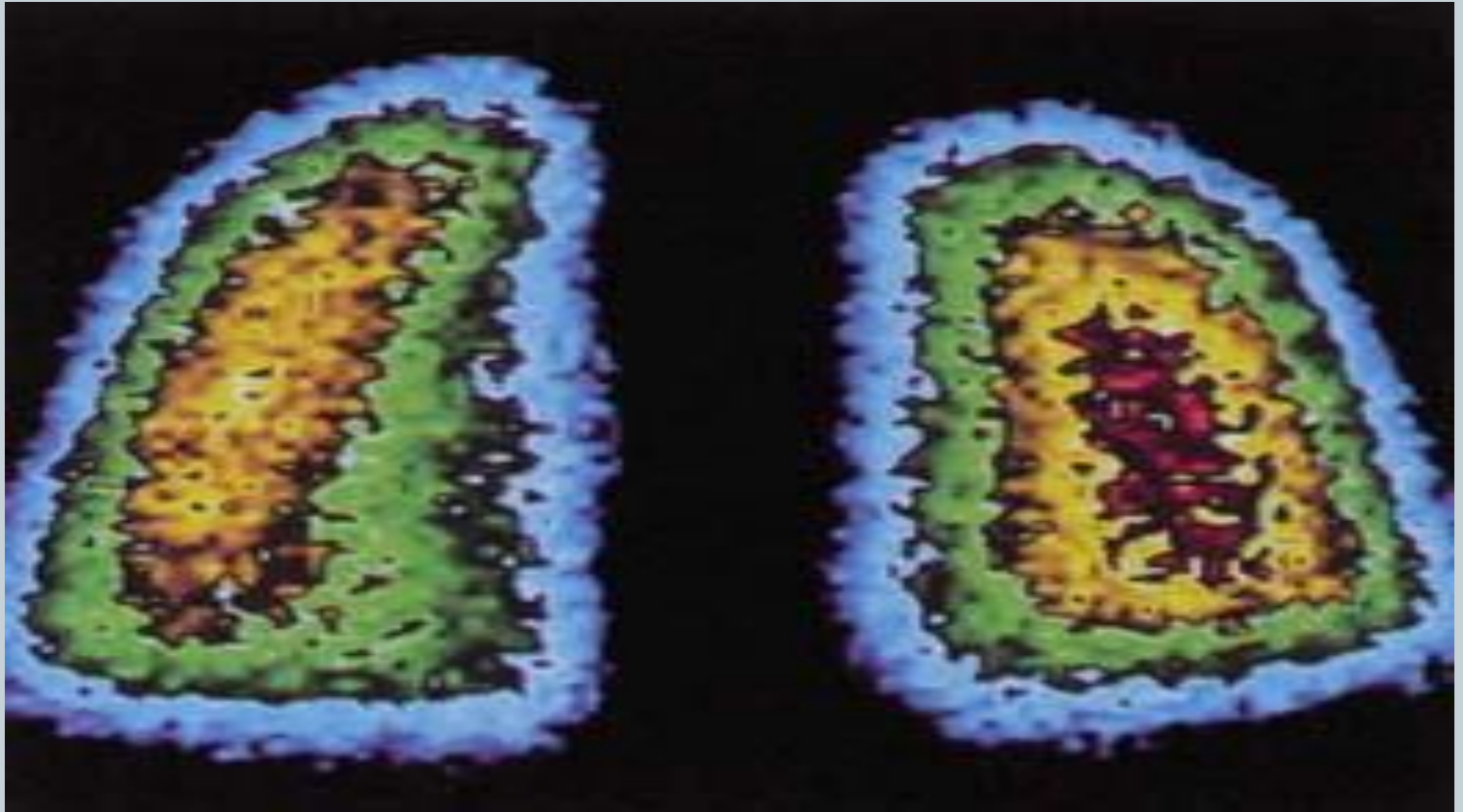


Image taken by PET to the left side ,the middle one taken by CT scan ,while the last one is both PET\CT scan combined image



**PET\CT scan image**

# Images of human lungs obtained from a $\gamma$ -ray scan



# MRI



Magnetic Resonance Image :  
Use Magnetic Field





# OPEN MRI



**VS**

# CLOSED MRI









**THE END**  
**THANK YOU**